Regional Carbon Initiative for CO₂ Management in Basalt: *CaRBTAP*

Richard Middleton | CARBON SOLUTIONS Houston CCUS, 3–5 March 2025





DOE Initiative

- Regional Initiative for Technical Assistance Partnerships (RITAP).
- Accelerate the understanding of specific geologic basins to enable the permanent storage of CO₂.

#	Basin	Prime
1	Appalachian Basin	Battelle
2	Central Valley	Stanford
3	Columbia River Basalt Group	Carbon Solutions
4	Permian Basin	Texas Tech University
5	Williston Basin	University of North Dakota
6	NW Gulf of Mexico Basin	The University of Texas at Austin
7	Uinta Basin	The University of Utah
8	Green River and Wind River Basins	University of Wyoming
9	Michigan Basin	Western Michigan University

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Regional Carbon Initiative for CO₂ Management in Basalt: *CaRBTAP*

Research team

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Industry Board

Carbfix | CRC/TerraVault | Cella | Addis Energy...



<u>Why</u> *CaRBTAP?*

- Storage potential of basalt.
- 2. Hard-todecarbonize region.
- 3. Motivated region.
- 4. Multiple CO₂ projects.







What CaRBTAP?

- 1. Geologic Challenges.
- 2. Carbon Management Infrastructure.
- 3. Policy and Regulatory Framework.
- 4. Commercial Acceptance.

Task 1: Storage Resource Assessments

Focus

- Regional Geologic Characterization.
- Regional Reservoir Model.
- Regional Storage Potential Estimates.
- Regional Geologic Database
 Development.
- Cross-Cutting Opportunities.



SCO₂T^{PRO}: Nationwide saline storage potential



Task 2: Technical Assistance

Focus

- Project Development.
- Monitoring, Reporting, & Verification (MRV).
- Pore and Pressure Space Management and Optimization.
- Legal and Regulatory Assessments and Strategies.
- Carbon Transport, Planning, & Safety Assistance.



SimCCS^{PRO}: Industrial emissions & CCS infrastructure





CO₂ & Basalts 101

A primer

Why Basalts?

- Mineralization & storage.
- Highly reactive with supercritical CO₂.
- Self-sealing for leakage scenarios.
- Gigatonne-storage in the Columbia River.
- Common rock type with worldwide distribution.
- Few/no other co-located storage options.



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Gigaton commercial-scale carbon storage and mineralization potential in stacked Columbia River basalt reservoirs

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A B S T R A C T

Geologic carbon sequestration Supercritical carbon dioxide Basalts Mineralization Anticline Storage resources Stacked reservoirs

Keywords

ARTICLE INFO

This work presents a detailed supercritical CO₂ storage resource estimation for the stacked basalt reservoirs in the Grande Ronde Basalt of the Columbia River Basalt Group in eastern Washington and Oregon. The assessment aims to derisk the commercialization potential of geologic carbon storage in basalt by leveraging both structural and mineralization trapping of CO₂ in basalt. The structural closures formed by anticlinal ridges and synclinal valleys in Yakima Fold Belt are excellent physical traps to accommodate injected supercritical CO₂. Rigorous hydraulic testing, well logs and simulation results from the Wallula Basalt Pilot #1 well showed the courrence of 17 suitable permeable injection zones (up to 2,496 mD) intercalated with dense seals ($-2.6E\cdot10$ mD) in the Grand Ronde Basalt mineralis to form stable earbonates. Our calculation indicates up to 40 gigatons (P90) of mineralization storage resources exist in the Grande Basalt reservites.

1. Introduction

Columbia River Basalt Group (CRBG) extends over a large area and is coincident with a very large fraction of the power generation and industrial CO₂ sources in the Pacific Northwest (PNW). However, because flood basalts represent a significant barrier for oil and gas exploration in the underlying sedimentary rock, this region is sparsely investigated relative to conventional sedimentary basins. Only very limited data are available on deep reservoir hydrogeologic properties that are critical to any quantitative assessment of potential for CO₂ storage.

To address these data gaps, Pacific Northwest National Laboratory (PNRL) has pioneered both laboratory (McGrail et al., 2009a, 2006; Schaef et al., 2014; Schaef and McGrail, 2009, 2010; Schaef et al., 2011; Bartels, 2024; Xiong et al., 2018) and field pilot studies (Depp et al., 2022; McGrail et al., 2017, 2009b; Polites et al., 2022; White et al., 2020; axamining the potential for large-scale injection and storage of CO₂ and natural gas (Reidel et al., 2002) in flood basalts. The PNNL-led Wallula Basalt Carbon Storage Pilot Project demonstrated the successful injection and mineralization of 977 tons of supercritical CO₂ (scCO₂) in the interflow zones of the Grand Ronde Basalt (GRB) near the town of Wallula in eastern Washington State. The post-injection core samples and hydrologic simulation reported that up to 60 % of injected CO₂ had been incorporated into carbonate mineralis within two years of injection (White et al., 2020). The simulated mineralization rate was consistent with laboratory observation, thus offering a highly secure pathway to permanent geologic CO₂ storage.

With an expansion of the 45Q tax credit in 2022, the PNW has an unprecedented opportunity to leverage a groundswell of interest and enthusiasm from potential industry partners. Based on the pilot-scale field tests, simulation modeling, and geochemical analysis, this paper provides a robust CO₂ pore space storage resource estimation for anticlinal basalt reservoirs. The storage resource estimation laid the foundation for future deployment of commercial-scale carbon capture (such as point source or direct air capture (DAC)) at favorable locations in the PNW and characterization approaches necessary to support successful application for the U.S. Environmental Protection Agency Class VI wells

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Setting the Stage for CO₂ Mineralization

Mineralization settings

- In situ mineralization:
 - Porous reservoirs (e.g., basalts).
 - Fractured reservoirs (e.g., peridotite, serpentinite).
 - Hybrid systems (e.g., fractured basalt-hosted geothermal reservoirs, basalt-rich sandstones).
- Ex situ mineralization.
 - Mine tailings, soil amendments, & engineered systems that leverage non-ambient conditions.

Basalt chemistry

- Crystalline minerals (feldspars, pyroxenes, olivine) within a highly reactive glassy matrix
- Carbonate type controlled by depth, temperature, surface area, pre-existing secondary minerals, pressure, & water chemistry.





In Situ Geologic Storage of CO₂ in Basalts

- Illustration of the depth of geologic storage of carbon dioxide in a layered basalt.
- Mafic-ultramafic reservoirs occur 800 m below surface.
- As a reference, the Eiffel Tower (300 m) is shown alongside for scale.





Pure Phase vs. Dissolved Phase

Pure phase

- Free-phase supercritical CO₂ (scCO₂).
- 800 m & deeper to remain scCO₂.
- High-storage density, no water use.



Dissolved phase

- CO₂ dissolved in water (~1:25).
- Suitable above 800 m.
- Low-storage density, high water use.



Injection Strategies: Pure vs. Dissolved

- Pure- CO_2 vs. Aqueous- CO_2 : End members on a spectrum.
- Hybrid approaches to storage efficiency & sustainability.
- Customization: water content & optimizing carbonation efficiency.
- Engineered additives to enhance mineralization.
- Multiphase injection strategies with drilling technologies & approaches.





Basalts tested in the field

Project: Wallula Basalt Carbon Sequestration Pilot

- Pacific Northwest National Laboratory (PNNL).
- Test characterization well: (2009).
- Injection permit issued (2011).
- Extended hydraulic test characterization (2012).
- Injection: ~1,000 tonnes of CO₂ (2013).
- Final characterization/decommissioning (2015).
- Sidewall core characterization (2017).
- Reservoir Simulations (2020).





Basalts tested in the lab

Wallula findings

Relative Intensity

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ARBON

Solutions

25

30

- 50 sidewall cores (SWC).
- Carbonate nodules observed.
- XRD of nodule material identified ankerite.
- Isotopic signature confirmed the injected CO_2 was mineralized.

XRD Pattern for

35

°**2**⊖

unreacted calcite vein

XRD Pattern for isolated

40

carbonate nodule

-10

-20

-30

Ankerite Nodules

Post-Injected

5¹⁸Oxygen,‰

50

PDF file 05-0586. Calcite

PDF file 33-0282, Ankerit

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Field to Simulations

- Reservoir simulations: history matching and forecasting
 - Data representing before, during, and after injection.
 - Simulations for ground-truthing CO₂ mineralization.
- Rates and mechanisms of mineralization in multiphase CO₂-H₂O fluids
 - Mineral-specific reactive surface areas.
 - Upscaling reactive MD to derive rates.
 - Parameterizing wet CO₂ in reactive transport codes.
- Holistic understanding of CO₂ mineralization for all divalent cations, Al, Si, Fe3+, etc.
 - Net porosity/permeability generated during reactive crystallization, or only maintenance.
 - No measurable decreases in porosity, permeability, or injectivity at Wallula.







Take home message

Photo: Andrea Starr (PNNL



- Vast storage potential in the US & globe, particularly in underserved areas.
- Mineralization & conventional storage.
- Risk minimization, storage paradigm.
- Field, lab, & simulation testing.

CaRBTAP

- Motivated region, for energy solutions.
- Industry & government investments.
- Experienced CaRBTAP team.
- Roadmap for commercial success.



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Additional material

Phase Behavior of CO₂-H₂O Mixtures

- Up to ~5 mol% H₂O solubility in CO₂ at prototypical carbon storage conditions.
- Water-bearing CO_2 is more reactive than aqueous-dissolved CO_2 .
- а 150 mol% solubility of H_2O in $CO_{2} = 0.01$ 0.1 125 10 100 scCO Pressure (bar) 100 0.1 0.2 75 0.3 0.4 0.5 50 0.6 Gas 0.7 0.8 25 0.9 25 75 50 125 100 150 ARBON Temperature (°C)

Solutions



- Nanoscale water films control unique mechanisms and anomalous energy barriers.
- Accessing new carbon mineralization routes.



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CO₂ Mineralization Reaction Pathway Endpoints

- CO₂ mineralization processes are overlapping and complex.
- Observed carbon mineralization assemblages at Wallula include aragonite (CaCO₃), siderite (FeCO₃), ankerite [Ca(Fe,Mn)(CO₃)2], and non-carbonate minerals.
- Chemically-zoned carbonate nodules (Ca, Mn, & Fe).
- Unique chemistries directly correlated to pre-existing pore-lining minerals.



Depp et al., 2022, Pore-scale Microenvironments Control Anthropogenic Carbon Mineralization Outcomes in Basalt, ACS Earth Space Chem



Polites et al. 2022, Exotic Carbonate Mineralization Recovered from a Deep Basalt Carbon Storage Demonstration, ES&T



Lahiri et al., 2023, Facile Metal Release from Pore-lining Phases Enables Unique Carbonate Zonation in a Basalt Carbon Mineralization Demonstration, ES&T

Reservoir Simulator Parameterization

- Mg-deficient, ordered ankerite structure.
- Rim region expanded siderite structure.
- Differential phase contrast imaging techniques further confirm the structure of the carbonate.
- No synthetic or natural ankerite ever observed.





Expanded Ca-rich Siderite structure

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